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Mark Davison, Roger A. Lalancette* and Hugh W. Thompson

Carl A. Olson Memorial Laboratories, Department of Chemistry, Rutgers University, Newark, NJ 0712, USA

Correspondence e-mail: rogerlal@andromeda.rutgers.edu

Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(\text{C}-\text{C}) = 0.004 \text{ Å}$ R factor = 0.055 wR factor = 0.162 Data-to-parameter ratio = 15.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Formation of non-centrosymmetric dimers in (1*S**,4a*R**,8a*R**)-decahydronaphthalene-1-carboxylic acid

The all-*cis* racemate of the title compound, $C_{11}H_{18}O_2$, crystallizes in the chair-chair conformation that places the carboxyl group in an equatorial postion. The space group is centrosymmetric but the compound aggregates as dimers whose components are related by a C_2 axis [$O \cdot \cdot \cdot O = 2.665$ (3) Å and $O - H \cdot \cdot \cdot O = 177^{\circ}$]. In the crystal structure, one $C - H \cdot \cdot \cdot O = C$ close contact is found.

Comment

The title compound, (I), is a racemic mixture of the 1S,4aR,8aR diastereomer and its enantiomer. The molecular structure of (I) is illustrated in Fig. 1. The carboxyl group, attached in an equatorial position in the expected double-chair conformation, is found to be fully ordered and lies approximately parallel to the C1–C2 bond [O1–C9–C1–C2 = -13.1 (3)°], so avoiding interaction with the C8 equatorial H atom.



Fig. 2 shows the dimeric carboxyl unit, whose components are related by a twofold axis normal to the carboxyl plane. Such non-centrosymmetric dimers arise most frequently in chiral non-racemates, where no centrosymmetry is possible (Gavezzotti & Filippini, 1994; Allen et al., 1999; Sørensen & Larsen, 2003). Our own study of the crystallography of ketocarboxylic acids has encountered several instances among achiral compounds and racemates of the formation of dimers lacking any element of symmetry (Lalancette et al., 1991, 1996), and one case of a C2-dimer (Coté et al., 1996). However, compound (I) is the only instance among our compounds of C_2 -dimer formation in a racemic acid. Within the 2.7 Å range we survey for non-bonded $C-H \cdots O$ packing interactions (Steiner, 1997), a pair of 2.70 Å close contacts is found, to the C=O from atom H2A in molecules centrosymmetrically related across $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ and the *c* cell edge.

Fig. 3 shows the crystal packing with extracellular molecules included, and shows the centrosymmetric packing of the non-

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A view of the asymmetric unit of (I), with the numbering scheme. Displacement ellipsoids are drawn at the 20% probability level.



Figure 2

A view of the carboxyl dimer, pairing molecules of identical handedness related by a C_2 axis normal to the carboxyl plane. Displacement ellipsoids are drawn at the 20% probability level. The $O-H \cdots O$ hydrogen bonds are shown as dashed lines.

centrosymmetric dimer units. The C_2 axes of the dimers lie parallel to b.

Experimental

Exhaustive catalytic hydrogenation of 6-hydroxy-1-naphthoic acid dissolved in 95% ethanol over a 5% Rh/C catalyst yielded no isolable hydroxy acid, but gave compound (I) as the major product (Ungnade & Morriss, 1950). The crystal used was obtained from acetonitrile (m.p. 398 K).

Crystal data

$C_{11}H_{18}O_2$	$D_x = 1.168 \text{ Mg m}^{-3}$
$M_r = 182.25$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 24
a = 21.893 (9) Å	reflections
b = 7.756 (3) Å	$\theta = 3.5 - 8.9^{\circ}$
c = 13.187 (7) Å	$\mu = 0.08~\mathrm{mm}^{-1}$
$\beta = 112.23 \ (3)^{\circ}$	T = 296 (2) K
$V = 2072.7 (16) \text{ Å}^3$	Parallelepiped, colourless
Z = 8	$0.50 \times 0.47 \times 0.15 \text{ mm}$





A partial packing diagram for (I), showing the non-centrosymmetric dimers arranged centrosymmetrically around $\left(\frac{1}{2727}\right)$ and the *c* cell edge. Displacement ellipsoids are drawn at the 20% probability level. The O-H...O hydrogen bonds are shown as dashed lines. For clarity, all C-bound H atoms have been removed.

Data collection

Siemens P4 diffractometer	$R_{\rm int} = 0.026$
ω –2 θ scans	$\theta_{\rm max} = 25.0^{\circ}$
Absorption correction: analytical	$h = -1 \rightarrow 26$
(SHELXTL; Sheldrick, 1997b)	$k = -9 \rightarrow 1$
$T_{\min} = 0.957, \ T_{\max} = 0.987$	$l = -15 \rightarrow 14$
2324 measured reflections	3 standard reflections
1840 independent reflections	every 97 reflections
1247 reflections with $I > 2\sigma(I)$	intensity decay: variation <1.9%
Refinement	

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0732P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.055$	+ 1.5893P]
$wR(F^2) = 0.162$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.01	$(\Delta/\sigma)_{\rm max} < 0.001$
1840 reflections	$\Delta \rho_{\rm max} = 0.49 \ {\rm e} \ {\rm \AA}^{-3}$
119 parameters	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
	Extinction coefficient: 0.0037 (10)

All H atoms for (I) were found in electron-density difference maps, but were placed in calculated positions and allowed to refine as riding atoms [C-H = 0.98 Å for methine groups and C-H = 0.97 Åfor methylene groups, with $U_{iso}(H) = 1.2U_{eq}(C)$, and O-H = 0.82 Å, with $U_{iso}(H) = 1.5U_{eq}(O)$].

Data collection: XSCANS (Siemens, 1996); cell refinement: XSCANS; data reduction: XSCANS; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine

structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXL97*.

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References

Allen, F. H., Motherwell, W. D. S., Raithby, P. R., Shields, G. P. & Taylor, R. (1999). New J. Chem. 23, 25–34.

- Coté, M. L., Thompson, H. W., Lalancette, R. A. & Williams, J. A. IV (1996). Acta Cryst. C52, 2612–2614.
- Gavezzotti, A. & Filippini, G. (1994). J. Phys. Chem. 98, 4831-4837.
- Lalancette, R. A., Coté, M. L. & Thompson, H. W. (1996). Acta Cryst. C52, 244–246.
- Lalancette, R. A., Thompson, H. W. & Vanderhoff, P. A. (1991). Acta Cryst. C47, 986–990.
- Sheldrick, G. M. (1997a). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). XSCANS. Version 2.2. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sørensen, H. O. & Larsen, S. (2003). Acta Cryst. B59, 132-140.
- Steiner, T. (1997). Chem. Commun. pp. 727-734.
- Ungnade, H. E. & Morriss, F. V. (1950). J. Am. Chem. Soc. 72, 2112-2115.